Oscillatory Rarefied Gas Flows in Long Capillaries

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Abstract Oscillatory, rarefied, linear and nonlinear fully developed flows of single 5 gases and binary gas mixtures, driven by external harmonic mechanisms with arbi- 6 trary frequency, have been recently considered by the authors in a series of works. 7 Here, these works are reviewed by focusing on the most notable findings. More 8 specifically, the effects of the oscillation frequency on the velocity overshooting 9 and gas separation phenomena in gas mixture flows and of the oscillation amplitude 10 on the flow pattern in nonlinear single gas flows are presented. Modeling is based 11 in the former case on the McCormack kinetic model and in the latter one on the 12 DSMC method. In general, as the flow becomes more rarefied higher frequencies are 13 needed to trigger the overshooting phenomenon, which becomes more pronounced 14 as the molecular mass of the gas species is increased. Notably, gas separation may 15 be present in the whole range of gas rarefaction, provided that the flow is subject 16 to adequate high oscillation frequency. Finally, the presence of strong external 17 harmonic forces does not significantly affect the oscillatory macroscopic quantities, 18 including the mass flow rate (no distortion of the amplitude-frequency curve), except 19 of the oscillatory axial heat flux, which exhibits a non-sinusoidal pattern. 20

1 Introduction

Rarefied boundary-driven oscillatory flows of single gases have been extensively ²² investigated over the last two decades [1–7]. These flows are present in various resonator structures [8, 9], while acoustic enhancement or attenuation (even cloaking) ²⁴ may be achieved in viscous-thermal fluids [10]. Propagation of sound waves due ²⁵ to mechanical and thermal excitation through binary gas mixtures has been also ²⁶ considered [11–13]. ²⁷

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The corresponding rarefied pressure-driven oscillatory gas flows have attracted ²⁸ much less attention, although there are employed in vapor deposition [14], microflu-²⁹ idic oscillators and pumps [15] and cryogenic pulse tubes [16]. Of course, in the ³⁰ hydrodynamic regime, pressure-driven oscillatory gas flows have been thoroughly ³¹ examined and are encountered in numerous technological fields ranging from ³² pneumatic lines and control systems [17], reciprocating pumps [18], combustion ³³ engines, and bioengineering to enhancement of thermal diffusion in mass and heat ³⁴ transfer processes, species contaminants dispersion and gas separation or mixing ³⁵ [19, 20]. Experimentally, oscillatory-type pressure-driven gas flows may be realized ³⁶ by reciprocating pistons [21] or membranes [22] or by oscillating the channel itself ³⁷ [23].

Although boundary and pressure gradient oscillatory flows have certain similarities, such as the traveling wave disturbance causing the flow, they also have various 40 differences related to the involved physical phenomena and quantities of practical 41 interest. The general mechanisms occurring in oscillatory boundary-driven flows 42 include inertia and viscous forces, while in pressure gradient flows, in addition to the 43 above, pressure forces are also considered. In the latter case, the difference in time 44 scales of pressure and viscous forces may lead to unexpected results, such as the 45 annular effect (velocity overshooting) and enhanced gas separation, which are not 46 observed in former case. Also, in boundary-driven flows we are mainly interested 47 in velocity and shear stresses, while in oscillatory pressure gradient flows including 48 pulsatile flows, we are also interested in the computed flow rates. 49

Taking into consideration that oscillatory pressure- driven gas flows in the ⁵⁰ hydrodynamic regime are very common, along with the progress in fabrication ⁵¹ techniques of micro devices, it is reasonable to expect that oscillatory pressure- ⁵² driven rarefied flows of single gases and gas mixtures will be also widely employed, ⁵³ in the short future. Therefore, very recently, some theoretical studies in fully ⁵⁴ developed oscillatory gas flows in capillaries [24–27] have been reported. Here, ⁵⁵ the most notable results of the detailed analysis in [24–27] for linear and nonlinear ⁵⁶ fully developed flows of single gases and binary gas mixtures are presented. ⁵⁷

2 Linear Oscillatory Fully Developed Binary Gas Mixture Flow

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Consider the time-dependent, isothermal, rarefied flow of a binary gas mixture 60 between two infinite long parallel plates fixed at $y' = \pm H/2$, connecting two 61 containers, as shown in Fig. 1. The pressure in the two containers harmonically 62 oscillates as $\tilde{P}_j(t') = \mathbb{R}[P_j \exp(-i\omega t')], j = 1, 2$, resulting in the externally



Fig. 1 Oscillatory flow configuration

imposed harmonically oscillating pressure gradient, along the parallel plates, of the 63 form 64

$$\frac{d\tilde{P}}{dx'} = \mathbb{R}\left[\frac{dP}{dx'}\exp\left(-i\omega t'\right)\right].$$
(1)

Here, $\tilde{P}(x',t') = P(x') \exp(-i\omega t')$ is the oscillatory pressure in the x'-65 direction parallel to the plates, dP/dx' and ω refer to the amplitude and frequency, 66 respectively, of the oscillatory pressure gradient $d\tilde{P}/dx'$ and t' is the time, while 67 \mathbb{R} denotes the real part of a complex expression $i = \sqrt{-1}$). The well-established 68 assumption that the fluid oscillates in bulk or en mass, i.e., that all quantities oscillate 69 with the same frequency as the pressure gradient, is applied [28]. Thus, this is an 70 harmonically oscillating, fully developed flow (pressure and density remain constant 71 at each cross section, while all other macroscopic distributions depend only in the 72 y'-direction normal to the plates).

The binary gas mixture consists of two monatomic species of molecular masses 74 m_{α} , with the index " $\alpha = 1, 2$," always referring, without loss of generality, 75 to the light and heavy species of the mixture, respectively. The corresponding 76 local number densities of the mixture components, defined by $\tilde{n}_{\alpha}(t')$, oscillate 77 harmonically as $\tilde{n}_{\alpha}(t') = \mathbb{R}[n_{\alpha} \exp(-i\omega t')]$, where $n_{\alpha}, \alpha = 1, 2$, is the local 78 amplitude of the oscillating number density of each species. The number density 79 of the mixture is $\tilde{n}(t') = \tilde{n}_1(t') + \tilde{n}_2(t')$, while the molar fraction of the mixture so is defined as the ratio of the number density of the light species over the mixture 81 number density, given by $\tilde{C}(t') = \mathbb{R}[C \exp(-i\omega t')]$, with $C = n_1/n = 82$ $n_1/(n_1 + n_2)$, being the local amplitude of the molar fraction. The molar fraction 83 amplitude of the heavy species is 1 - C. The mean molecular mass of the mixture ⁸⁴ is given by $m = Cm_1 + (1 - C)m_2$. The number densities of the species and 85 the mixture are related to the corresponding pressures with the equation of states 86 as $\tilde{P}_{\alpha} = \tilde{n}_{\alpha}kT$ and $\tilde{P} = \tilde{n}kT$, respectively, where \tilde{P}_{α} are the partial pressures, 87 $\tilde{P} = \tilde{P}_1 + \tilde{P}_2$ is the total pressure, T is the reference temperature. The mass densities 88 of the species and the mixture are defined as $\rho_{\alpha} = m_{\alpha}n_{\alpha}$ and $\rho = mn$, respectively. 89

The deduced time-dependent flow quantities of practical interest include the bulk 90 velocity $\tilde{U}_{\alpha}(t', y')$, shear stress $\tilde{\Pi}_{\alpha}(t', y')$ and heat flow $\tilde{Q}_{\alpha}(t', y')$ of the two 91

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species $\alpha = 1, 2$, which depend on y', the space independent variable vertical to 92 the plates and vary harmonically with time t' as 93

$$\tilde{Z}_{\alpha}\left(t', y'\right) = \mathbb{R}\left[Z_{\alpha}\left(y'\right)\exp\left(-i\omega t'\right)\right],\tag{2}$$

where $\tilde{Z}_{\alpha}(t', y') = \left[\tilde{U}_{\alpha}(t', y'), \tilde{\Pi}_{\alpha}(t', y'), \tilde{Q}_{\alpha}(t', y')\right]$, while $\tilde{Z}_{\alpha}(y') = _{94}$ $\left[U_{\alpha}(y'), \Pi_{\alpha}(y'), Q_{\alpha}(y')\right]$ is a vector of the corresponding complex functions. ⁹⁵ In addition, the oscillatory particle flow rates of the two species are given by ⁹⁶ $\tilde{J}_{\alpha}(t') = \mathbb{R}\left[J_{\alpha}\exp\left(-i\omega t'\right)\right]$, where $J_{\alpha} = n_{\alpha}\int_{\alpha}^{H/2}U_{\alpha}dy'$, as well as the ⁹⁷

corresponding mixture particle flow rate $\tilde{J} = \tilde{J}_1 + \tilde{J}_2$, are complex functions.

Furthermore, the dimensionless independent space and time variables x = 99x'/H, y = y'/H and $t = t'\omega$, are introduced. The dimensionless amplitude of the 100 local oscillatory pressure gradient is 101

$$X = \frac{H}{P(x')} \frac{dP(x')}{dx'} = \frac{1}{P(x)} \frac{dP(x)}{dx} \ll 1.$$
 (3)

The bulk velocity, shear stress and heat flow in Eq. (2) are nondimensionalized by 102 (υX) , (2PX) and (υPX) , respectively, with $\upsilon = \sqrt{2kT/m}$ being the characteristic 103 speed of the mixture, to yield: 104

$$\tilde{\varphi}_{\alpha}(t, y) = \mathbb{R}\left[\varphi_{\alpha}(y) \exp\left(-it\right)\right] = \varphi_{\alpha}^{(A)}(y) \cos\left[t - \varphi_{a}^{(P)}(y)\right],\tag{4}$$

where $\tilde{\varphi}_{\alpha}(t, y) = [\tilde{u}_{\alpha}(t, y), \tilde{\varpi}_{\alpha}(t, y), \tilde{q}_{\alpha}(t, y)]$. In Eq. (4) the superscripts (A) 105 and (P) refer to the amplitude and the phase angle, respectively, of each complex 106 quantity. 107

Furthermore, the flow rates $\tilde{J}_{\alpha}(t')$ are nondimensionalized by $(PXH/m\upsilon)$ to 108 obtain the dimensionless oscillatory particle flow rates of each species 109

$$\tilde{G}_{\alpha}(t) = \mathbb{R}\left[G_{\alpha}\exp\left(-it\right)\right] = \mathbb{R}\left[G_{\alpha}^{(A)}\exp\left[i\left(G_{\alpha}^{(P)}-t\right)\right]\right] = G_{\alpha}^{(A)}\cos\left[t-G_{\alpha}^{(P)}\right],$$
(5)

where

$$G_{\alpha} = G_{\alpha}^{(A)} \exp\left(iG_{\alpha}^{(P)}\right) = 2\int_{-1/2}^{1/2} u_{\alpha}dy.$$
 (6)

Also, the dimensionless oscillatory particle flow rate of the mixture is given by

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$$\tilde{G}(t) = \mathbb{R}\left[G\exp\left(-it\right)\right] = \mathbb{R}\left[G^{(A)}\exp\left[i\left(G^{(P)}-t\right)\right]\right] = G^{(A)}\cos\left[t-G^{(P)}\right],\tag{7}$$

where $G = CG_1 + (1 - C) G_2$, with the superscripts (A) and (P), always referring 112 to amplitudes and phase angles, respectively. 113

The oscillatory binary gas mixture flow between parallel plates is also characterized by the gas rarefaction and oscillation parameters, given by

$$\delta = \frac{PH}{\mu\upsilon}$$
 and $\theta = \frac{P}{\mu\omega}$, (8)

respectively, where μ is the gas viscosity at some reference temperature T, v is the 116 characteristic speed of the mixture, the ratio (P/μ) is the intermolecular collision 117 frequency. The composition of the binary gas mixture, i.e., the molecular masses m_1 118 and m_2 , as well as the amplitude of the molar fraction C, must be also specified. 119

Next, the kinetic formulation, based on the McCormack model [29], is shortly 120 presented. Due to the condition $X \ll 1$ the unknown time-dependent distribution 121 function of each species can be linearized in a standard manner and the linearized 122 distributions are accordingly projected to yield the following set of kinetic equations: 124

$$-i\frac{\delta}{\theta}\sqrt{\frac{m_{\alpha}}{m}}\Phi_{\alpha} + c_{\alpha y}\frac{\partial\Phi_{\alpha}}{\partial y} + \omega_{a}\boldsymbol{\gamma}_{\alpha}\Phi_{a} =$$

$$-\frac{1}{2}\sqrt{\frac{m}{m_{a}}} + \omega_{\alpha}\left\{\boldsymbol{\gamma}_{\alpha}\boldsymbol{u}_{a} - \boldsymbol{v}_{\alpha\beta}^{(1)}\left(\boldsymbol{u}_{a} - \boldsymbol{u}_{\beta}\right) - \frac{1}{2}\boldsymbol{v}_{\alpha\beta}^{(2)}\left(\boldsymbol{q}_{a} - \frac{m_{a}}{m_{\beta}}\boldsymbol{q}_{\beta}\right) + \frac{125}{2\sqrt{\frac{m}{m_{a}}}}\left[\left(\boldsymbol{\gamma}_{\alpha} - \boldsymbol{v}_{\alpha\alpha}^{(3)} + \boldsymbol{v}_{\alpha\alpha}^{(4)} - \boldsymbol{v}_{\alpha\beta}^{(3)}\right)\boldsymbol{\varpi}_{a} + \boldsymbol{v}_{\alpha\alpha}^{(4)}\boldsymbol{\varpi}_{\beta}\right]c_{ay} +$$

$$126$$

$$+\frac{2}{5}\left[\left(\boldsymbol{\gamma}_{\alpha}-v_{\alpha\alpha}^{(5)}+v_{\alpha\alpha}^{(6)}-v_{\alpha\beta}^{(5)}\right)q_{a}+v_{\alpha\beta}^{(6)}\sqrt{\frac{m_{\beta}}{m_{a}}}q_{\beta}-\frac{5}{4}v_{\alpha\beta}^{(2)}\left(u_{a}-u_{\beta}\right)\right]\left(c_{ay}^{2}-\frac{1}{2}\right)\right\},$$
(9)

$$-i\sqrt{\frac{m_{\alpha}}{m}}\frac{\delta}{\theta}\Psi_{\alpha}+c_{\alpha y}\frac{\partial\Psi_{\alpha}}{\partial y}+\omega_{\alpha}\boldsymbol{\gamma}_{\alpha}\Psi_{a}=$$

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$$=\frac{4}{5}\omega_{\alpha}\left[\left(\boldsymbol{\gamma}_{\alpha}-\boldsymbol{v}_{\alpha\alpha}^{(5)}+\boldsymbol{v}_{\alpha\alpha}^{(6)}-\boldsymbol{v}_{\alpha\beta}^{(5)}\right)q_{a}+\boldsymbol{v}_{\alpha\beta}^{(6)}\sqrt{\frac{m_{\beta}}{m_{a}}}q_{\beta}-\frac{5}{4}\boldsymbol{v}_{\alpha\beta}^{(2)}\left(\boldsymbol{u}_{a}-\boldsymbol{u}_{\beta}\right)\right].$$
 (10)

Here, Φ_a and Ψ_a are complex perturbed distribution functions for each species, 130 $\omega_{\alpha} = \delta \left(C/\gamma_1 + (1-C)/\gamma_2 \right) \sqrt{m_a/m}$ and γ_a (a = 1, 2) are the collision 131 frequencies of each species [30]. Also, $\alpha, \beta = 1, 2$, with $\alpha \neq \beta$, while the 132

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expressions for the quantities $v_{\alpha\beta}^{(k)}$ are given in terms of the Chapman-Cowling 133 integrals as in [30]. The macroscopic quantities u_{α} , ϖ_{α} and q_{α} at the right hand 134 side of Eqs. (9) and (10) are defined in Eq. (4), respectively, and after applying the 135 linearization and projection procedures, they are obtained as moments of Φ_{α} and 136 Ψ_{α} as follows: 137

$$u_{\alpha}(y) = \frac{1}{\sqrt{\pi}} \int_{-\infty}^{\infty} \Phi_a \exp\left(-c_{ay}^2\right) dc_{ay}, \qquad (11)$$

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$$\overline{\omega}_{\alpha}(y) = \frac{1}{\sqrt{\pi}} \sqrt{\frac{m_a}{m}} \int_{-\infty}^{\infty} \Phi_a c_{ay} \exp\left(-c_{ay}^2\right) dc_{ay}, \tag{12}$$

$$q_{\alpha}(y) = \frac{1}{\sqrt{\pi}} \int_{-\infty}^{\infty} \left[\Psi_a + \left(c_{ay}^2 - \frac{1}{2} \right) \Phi_a \right] \exp\left(-c_{ay}^2 \right) dc_{ay}.$$
(13)

In the present work purely diffuse reflection at the walls is assumed.

The above set of equations is computationally solved based on the discrete 141 velocity method in the c_y space and on the second-order diamond finite difference 142 scheme in the *y* space. The discretized equations are solved in an iterative manner 143 between the kinetic equations (9) and (10) and the moment equations (11)–(13). 144 More information about the numerical scheme may be found in [27]. 145

Computational results are presented for the mixture flow rate amplitude and 146 phase angle (Fig. 2), the velocity and shear stress distributions (Fig. 3) and the 147 ratio of the flow rate amplitudes of the species (Fig. 4) in a wide range of the 148 gas rarefaction and oscillation parameters δ and θ , as well as of the molar fraction 149 $C \in [0, 1]$ for the He–Xe mixture with $m_2/m_1 = 32.8$.

In Fig. 2, the He–Xe flow rate amplitude $G^{(A)}$ and phase angle $G^{(P)}$ are presented 151 in terms of δ , with $\theta = [1, 100]$ and C = [0, 0.5, 0.9]. The results with C = 152



Fig. 2 Mixture flow rate amplitude $G^{(A)}$ and phase angle $G^{(P)}$ of He-Xe vs δ



Fig. 3 Velocity $u_{\alpha}^{(A)}(y)$ and shear stress $\overline{\omega}_{\alpha}^{(A)}(y)$ amplitudes of each species of He-Xe for $\delta = 10$ and $\theta = 0.1$. Reprinted with permission from [27]. Copyright (2022) by the American Physical Society



Fig. 4 Ratio of flow rate amplitudes $G_1^{(A)}/G_2^{(A)}$ of the species of He-Xe vs $\delta \in [10^{-4}, 10^2]$. Reprinted with permission from [27]. Copyright (2022) by the American Physical Society

0 correspond to the oscillatory single gas flow reported in [25]. The flow rate 153 amplitudes and phase angles of the mixture ($C \neq 0$) depend on the flow parameters 154 similarly to the corresponding single gas ones (C = 0). Always, the mixture flow 155

rate amplitude is larger and the phase angle is smaller than the corresponding ones of 156 the single gas. At large θ the dependency of $G^{(A)}$ on δ , is not monotonic, indicating 157 that there is a critical δ to obtain the maximum flow rate, while at small θ , $G^{(A)}$ is 158 decreased monotonically. This is due to the fact that at low oscillation frequencies 159 and as long as $\delta \ll \theta$, the variation of $G^{(A)}$ with δ has some resemblance with the 160 steady one, including the presence of the Knudsen minimum. Then, as δ is further 161 increased the effect of the inertia forces becomes significant and $G^{(A)}$ is decreased. 162 In addition, as θ is decreased (the oscillation frequency is increased), $G^{(A)}$ is always 163 decreased, while $G^{(P)}$ (the phase angle lag with respect to the pressure gradient) is 164 always increased reaching the limiting value of $\pi/2$.

In Fig. 3, the distributions of the velocity and shear stress amplitudes $u_{\alpha}^{(A)}(y)$ and 166 $\overline{\omega}_{\alpha}^{(A)}(y)$ of each species of the He–Xe gas mixture, with C = [0.1, 0.4, 0.7, 0.9], 167 are provided for $\delta = 10$ and $\theta = 0.1$. The specific values of δ and θ are suitable 168 for investigating the velocity overshooting phenomenon in the light and heavy 169 species of the mixture. Velocity overshooting is due to the fact that close to the 170 wall, viscous and pressure gradient forces actually add to each other due to the 171 large phase angle lag between them. As a result, the combined effect accelerates the 172 fluid to higher velocities than those produced in the core by the pressure gradient 173 forces acting alone. For Xe, compared to He, the velocity overshooting becomes 174 sharper, appearing, along with its maximum value, closer to the wall inside a 175 much thinner layer. In the core of the flow, the velocity amplitudes of both He 176 and Xe become flat and they are close to the corresponding analytical amplitudes 177 $u_{\alpha}^{(A)} = (\theta/2\delta) (m/m_{\alpha})$ (see Section 3 in [27]). In parallel, $\overline{\omega}_{\alpha}^{(A)}(y)$ for both He 178 and Xe take their highest values at the wall and they are monotonically decreased 179 towards the channel center. The attenuation of the shear stress amplitude of He is 180 smooth, diffused in the whole distance from the wall to the center, while the one of 181 Xe is rapid in a narrow zone close to the wall and far from the wall the shear stress of 182 Xe becomes zero. Since the viscous forces in the case of He act in the whole distance 183 between the plates, while in the case of Xe only in thin zones close to the walls, 184 the above observations on the velocity overshooting of He and Xe are physically 185 justified. This description of the velocity and shear stress amplitudes remains valid 186 for all molar fractions tested [27]. In brief, it is seen that as the molecular mass 187 of the gas species increases, the species shear stress, which is created at the wall 188 and is diffused into the flow, attenuates more rapidly, i.e., the Stokes layer becomes 189 thinner and the Richardson effect more pronounced. Velocity overshooting may be 190 also present in even lower rarefaction parameters provided that higher oscillation 191 frequencies are applied [24]. 192

The gas separation phenomenon for various values of δ and θ is discussed next. ¹⁹³ Gas separation in rarefied steady-state pressure-driven binary gas flows though ¹⁹⁴ capillaries may be analyzed by computing the ratio of the particle flow rates J_1/J_2 , ¹⁹⁵ which is monotonically increased as δ is decreased up to its maximum value, equal ¹⁹⁶ to $\sqrt{m_2/m_1} (1 - C) / C$, in the free molecular limit ($\delta \rightarrow 0$) [31]. ¹⁹⁷

In Fig. 4, the ratio of the flow rate amplitudes $G_1^{(A)}/G_2^{(A)}$ is provided in terms of δ 198 for the He–Xe gas mixture, with C = [0, 05, 0.35, 0.65, 0.95] and $\theta = [0.1, 1, 10]$. 199

At $\theta = 10$ the ratio $G_1^{(A)}/G_2^{(A)}$ varies qualitatively similarly as in the steady- 200 state binary gas flow setup. It is about constant or slightly reduced in the free 201 molecular regime (at $\delta = 0$ it is equal to the corresponding steady one) and then 202 it is monotonically decreased asymptotically going in the slip and hydrodynamic 203 regimes to one. In the free molecular regime, with regard to the gas rarefaction 204 parameter, as $\delta \rightarrow 0$, with $\theta > 0$, Eqs. (9) and (10) tend to the corresponding 205 ones for steady-state binary gas flow in the free molecular limit [30]. However, at 206 $\theta = 1$ and $\theta = 0.1$ the behavior of $G_1^{(A)}/G_2^{(A)}$ is completely different. It remains 207 about constant in free molecular regime, but then, it is increased in the transition 208 regime and finally, as δ further increases, it keeps asymptotically increasing to some 209 constant value, which is the molecular mass ratio of the heavy over the light species 210 $m_2/m_1 (G_{He}^{(A)}/G_{Xe}^{(A)} = 32.8)$. This is in accordance to the closed-from expression 211 that as $\theta \to 0$, $G_1/G_2 = m_2/m_1$ [27]. This behavior, with the minimum and 212 maximum values of $G_1^{(A)}/G_2^{(A)}$ appearing at the free molecular and hydrodynamic 213 limits, respectively, and the increase in the transition regime (completely reversed 214 compared to the steady-state behavior) becomes more pronounced as θ is decreased. 215

It is evident that the oscillation parameter θ has a dominant effect on the 216 amplitude ratio of He over Xe, which is significantly increased as θ is decreased 217 (at $\theta = 0.1$ the flow rate amplitude of He is about thirty times larger than of 218 Xe). This behavior is due to the corresponding behavior of the velocity amplitudes 219 and it is contributed to inertia forces, which are increased with the oscillation 220 frequency and they influence the bulk velocity amplitude of the heavy species much 221 more than of the light one. Therefore, as θ is decreased, the flow rate amplitude 222 of the heavy species decreases much more significantly than the light one and 223 although both amplitudes are decreased the velocity amplitude ratio of the light 224 over the heavy species is increased. This effect is magnified as the flow becomes 225 less rarefied overcoming diffusion effects due to increased intermolecular collisions 226 and therefore, as δ increases the amplitude ratio keeps increasing. There is no 227 contradiction to general theory, since oscillatory flows approach the hydrodynamic 228 regime, only when both δ and θ are adequately large.

3 Nonlinear Oscillatory Fully Developed Single Gas Flow

Consider the oscillatory nonlinear fully developed flow of a monatomic rarefied 231 gas, confined between two parallel infinite plates at temperature T_0 located at 232 $y' = \pm H/2$, due to an external harmonic force acting on the gas per unit mass 233 in the *x*-direction parallel to the plates [26]. The external force is defined as 234 $\tilde{F}'(\omega, t') = F' \cos(\omega t')$, where F' is the force amplitude. The convenient 235 complex factor $\exp(-i\omega t')$ previously used, cannot be employed since the force 236 amplitude F' may be arbitrarily large and in nonlinear oscillatory flows the real and 237 imaginary parts are not separable.

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The oscillatory macroscopic distributions of practical interest, characterizing ²³⁹ the flow, include the *x*-component $U_{x'}(y', t')$ of the velocity vector, the number ²⁴⁰ density N(y', t'), the temperature T(y', t'), and the axial and normal heat flow ²⁴¹ components $Q_{x'}(y', t')$ and $Q_{y'}(y', t')$, respectively, with $-H/2 \le y' \le H/2$ and ²⁴² $0 \le t' \le 2\pi/\omega$. The most important overall quantities are the mass flow rate and ²⁴³ axial heat flow ²⁴⁴

$$M'(t') = m \int_{-H/2}^{H/2} N(y',t') U_{x'}(y',t') dy' \text{ and } \bar{Q}_{x'}(t') = \frac{1}{H} \int_{-H/2}^{H/2} Q_{x'} dy',$$
(14)

respectively, where *m* is the molecular mass.

The parameters defining the above dimensional flow setup include the rarefaction 246 parameter and oscillation parameter defined in Eq. (8). Also, the external force 247 parameter, defined as $F = F'H/v_0^2$, is needed. It is the inverse of the square of the 248 Froude number (*Fr*). The effect of the external force on the flow is increased with 249 *F* and nonlinear effects are becoming dominant. On the contrary, as *F* is decreased 250 the corresponding linear oscillatory flow, which is linearly proportional to the force 251 magnitude, is gradually recovered.

The following dimensionless variables are introduced:

$$x = \frac{x'}{H}, dx = \frac{dx'}{H}, y = \frac{y'}{H}, dy = \frac{dy'}{H}, t = \frac{t'}{(H/v_0)}$$
(15)

$$n = \frac{N}{N_0}, u_x = \frac{U_{x'}}{\upsilon_0}, \tau = \frac{T}{T_0}, p_{xy} = \frac{\Pi_{x'y'}}{2P_0}, p = \frac{P}{2P_0}, q_x = \frac{Q_{x'}}{\upsilon_0 P_0}, q_y = \frac{Q_{y'}}{\upsilon_0 P_0}.$$
(16)

The equation of state becomes $p = n\tau/2$.

Then, the dimensionless external force acting on the gas per unit mass becomes 256

$$\tilde{F}(\delta, \theta, t) = F \cos\left(\frac{\delta}{\theta}t\right),$$
(17)

while the dimensionless flow rate and axial heat flow are given by

$$M(t) = \frac{M'}{2P_0(H/\upsilon_0)} = \int_{-1/2}^{1/2} n(t, y) u(t, y) dy, \quad \bar{q}_x(t) = \int_{-1/2}^{1/2} q_x(y, t) dy.$$
(18)

Next, the typical DSMC approach, with the No Time Counter (NTC) scheme 258 proposed by Bird [32], is implemented. The time evolution of the particle system 259 within a small time interval $\Delta t'$ is split into two consecutive steps: free motion of all 260

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particles and binary collisions of particles. The time step $\Delta t'$ is nondimensionalized 261 as $\Delta t = \Delta t' / (H/v_0)$. Purely diffuse boundary conditions are considered at the 262 walls, while periodic boundary conditions are applied in the *x* – and *z* – directions. 263 Hard sphere (HS) molecules are assumed. The external force is introduced by 264 accordingly altering the particle velocities at each time step, during the free motion. 265

Numerical results of the dimensionless flow rate and axial heat flow are provided 266 in terms of the force amplitude F = [0.05, 0.1, 0.5], corresponding to small, 267 moderate, and large force amplitudes, in a wide range of δ and θ . Since the results 268 of the nonlinear gas flow are similar with the linear ones in terms of δ and θ , only 269 the effect of the force amplitude is here discussed. 270

In Fig. 5, the flow rate amplitudes G_A are divided by the external force F 271 in order to directly compare with the corresponding linear results (the linear 272 solution is proportional to F) and they are presented for $\delta = [0.1, 1, 10], \theta = 273$ $[0.1, 1, 10, 20, 10^2]$ and F = [0.05, 0.1, 0.5]. The linear flow rate amplitudes 274 obtained in [26] are also provided. It is seen that for F = 0.05 and F = 0.1 the 275 deviation between the corresponding DSMC and linear solutions is small for $\delta > 1_{276}$ and for all values of θ , while for $\delta = 0.1$ and $\theta = [10, 20, 10^2]$ the deviation 277 is increased. It is evident that nonlinear effects are becoming more pronounced in 278 highly rarefied atmospheres (small δ) and low frequencies (large θ). For F = 0.5 all 279 deviations between DSMC and linear results are further increased due to nonlinear 280 effects. Again, the largest deviations are occurring at $\delta = 0.1$ and $\theta = [10, 20, 10^2]$ 281 $(\delta << \theta)$, while the deviations remain small for $\delta > 1$, even at high frequencies. 282 Overall, it may be stated that the presence of strong external harmonic forces does 283 not significantly affect the mass flow rate of the oscillatory flow, i.e., there is no 284 distortion of the amplitude-frequency response curve. 285

The space-average axial flow $\bar{q}_x(t)$ is plotted over one cycle in Fig.6 for 286 $\delta = [0.1, 1, 10]$ and $\theta = [0.1, 1, 10, 20, 10^2]$ with F = [0.05, 0, 5]. It is readily 287 seen that there are significant qualitative differences between the corresponding 288 space-average heat flow for F = 0.05 and F = 0.5. For F = 0.05, $\bar{q}_x(t)$ for 289 all values of δ and θ has a sinusoidal behavior over time. For F = 0.5, $\bar{q}_x(t)$ 290 shows over one cycle various patterns. It is seen that for $\delta = 0.1$ with $\theta = 0.1$, 1] and for $\delta = 10$ and $\theta = [0.1, 1, 10]$, i.e., in all cases 292



Fig. 5 Normalized oscillatory flow rate amplitude G_A/F vs $\theta \in [10^{-1}, 10^2]$



Fig. 6 Space-average axial heat flux $\bar{q}_x(t)$ vs t with F = 0.05 (up) and F = 0.5 (down) and $\theta = [0.1, 1, 10, 20, 100]$. Reproduced from [26], with the permission of AIP Publishing

where $\delta \ge \theta$, $\bar{q}_x(t)$ exhibits a sinusoidal pattern. On the contrary, in all cases where 293 $\delta < \theta$, $\bar{q}_x(t)$ exhibits a rather complex non-sinusoidal pattern indicating that the 294 introduced nonlinearities are responsible for the generation of oscillatory motion 295 containing several harmonics. These results are in agreement with the discussion 296 in Fig. 5, where nonlinear effects are becoming more significant in highly rarefied 297 flow (small δ) and low oscillation frequencies (large θ). Also, for both values of 298 F, the amplitude of $\bar{q}_x(t)$, as of all other macroscopic quantities, is reduced with θ 299 and almost diminishes at very high frequencies, particularly as the gas becomes less 300 rarefied. 301

4 Concluding Remarks

A brief overview of rarefied, oscillatory, pressure-driven, linear and nonlinear, fully 303 developed flows of single gases and binary gas mixtures is provided, while the 304 detailed analysis may be found in [24–27]. Here, the discussion is focused on 305 the most notable findings, which include velocity overshooting, gas separation and 306 nonlinear effects. The following concluding remarks are stated: 307

 Velocity overshooting (or the so-called Richardson effect) is present in oscillatory, rarefied single and binary gas mixture flows, but as the flow becomes more rarefied higher frequencies are needed to trigger this phenomenon.

- Gas separation in oscillatory binary gas mixture, may be present in the whole 311 range of gas rarefaction provided that the flow is subject to adequate high 312 oscillation frequency. 313
- Range of applicability of linear theory is much wider than expected in terms ³¹⁴ of the imposed amplitude of the oscillatory pressure gradient. The oscillatory ³¹⁵ axial heat flux is the mostly affected quantity and the only one that, due to ³¹⁶ nonlinearities, may exhibit a complex pattern. ³¹⁷

The present results may be useful in the design of technological devices operating 318 at moderate and high frequencies in the whole range of gas rarefaction, applicable 319 in various technological fields. 320

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